

# Liquid Scoping Study for Tritium-Lean, Fast Ignition Inertial Fusion Energy Power Plants

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**August 14, 2001**

**U.S. Department of Energy**

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# **Liquid Scoping Study for Tritium-Lean, Fast Ignition Inertial Fusion Energy Power Plants\***

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## **Abstract**

In a thick-liquid protected chamber design, such as HYLIFE-II, a molten-salt is used to attenuate neutrons and protect the chamber structures from radiation damage. The molten-salt absorbs some of the material and energy given off by the target explosion. In the case of a fast ignition inertial fusion system, advanced targets have been proposed that may be self-sufficient in the tritium breeding (i.e., the amount of tritium bred in target exceeds the amount burned). These "tritium-lean" targets contain approximately 0.5% tritium and 99.5% deuterium, but require a large  $\rho r$  of 10-20 g/cm<sup>2</sup>. Although most of the yield is provided by D-T reactions, the majority of fusion reactions are D-D, which produces a net surplus of tritium. This aspect allows for greater freedom when selecting a liquid for the protective blanket (lithium-bearing compounds are not required).

This study assesses characteristics of many single, binary, and ternary molten-salts. Using the NIST Properties of Molten Salts Database, approximately 4300 molten-salts were included in the study [1]. As an initial screening, salts were evaluated for their safety and environmental (S&E) characteristics, which included an assessment of waste disposal rating, contact dose, and radioactive afterheat. Salts that passed the S&E criteria were then evaluated for neutron shielding ability and pumping power. The pumping power was calculated using three components: velocity head losses, frictional losses, and lift.

This assessment left us with 57 molten-salts to recommend for further analysis. Many of these molten-salts contain elements such as sodium, lithium, beryllium, boron, fluorine, and oxygen. Recommendations for further analysis are also made.

\* This work was performed under the auspices of the U.S. Department of Energy by the University of California, Lawrence Livermore National Laboratory under contract No. W-7405-Eng-48.

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## 1. Introduction

The idea of inertial fusion using fast-ignition has been proposed as a method of achieving relatively high gain using ultra-powerful lasers to ignite the fusion fuel [2]. Advanced targets have also been proposed that may be self-sufficient in the tritium breeding aspect. These “tritium-lean” targets contain approximately 0.5% tritium and 99.5% deuterium (see Fig. 1.1), but require a large pr of  $10\text{-}20\text{ g/cm}^2$  (compared to  $\sim 3\text{ g/cm}^2$  for conventional hot-spot ignition. About 55% of the energy released by S. Atzeni’s target is produced by D-T reactions, even though the majority (60.5%) of the reactions are D-D, which produces a new surplus of tritium [2,3,4]. Table 1.1 shows the tritium balance within a tritium-lean target [3]. Figure 1.3 shows the spectrum from Atzeni’s “tritium-lean” fusion target. In order to achieve 1 GW power plant output, and because of the large yield (1330 MJ), these targets will be ignited at a frequency of 1.7 Hz. The low repetition keeps the pumping power significantly lower than in a traditional 5-10 Hz system. These targets may be direct or indirectly driven, and Figure 1.2 shows a schematic of a potential setup for both systems.

Tritium Balance Within a Tritium-Lean Target	
• Initial tritium loading = 0.53% in 19.6 mg	
	= 156 $\mu\text{g}$
	= $3.11 \times 10^{19}$ tritons
• Fusion reactions:	
# of D-D fusions	= $5.01 \times 10^{20}$ per target ( $2.51 \times 10^{20}$ tritons created)
# of D-T fusions	= $2.58 \times 10^{20}$ per target ( $2.58 \times 10^{20}$ tritons destroyed)
# of D- $^3\text{He}$ fusions	= $8.68 \times 10^{19}$ per target
• Transmutation reactions:	
D(n, $\gamma$ )T	= $4.85 \times 10^{16}$ tritons created $\rightarrow$ 80% immediately burn
T(n, $2n$ )D	= $2.45 \times 10^{17}$ tritons destroyed
$^3\text{He}(n,p)\text{T}$	= $1.92 \times 10^{19}$ tritons created
• Overall tritium breeding:	
Initial tritium	= $3.11 \times 10^{19}$ tritons/target
Final tritium	= $4.31 \times 10^{19}$ tritons/target
<u>Tritium-lean target breeding ratio</u> = 1.38	

Table 1.1 Tritium-balance for tritium-lean target [3].

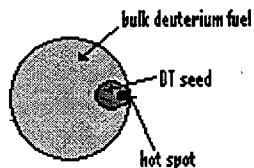


Figure 1.1 Sketch of a typical tritium-lean fuel assembly [4].

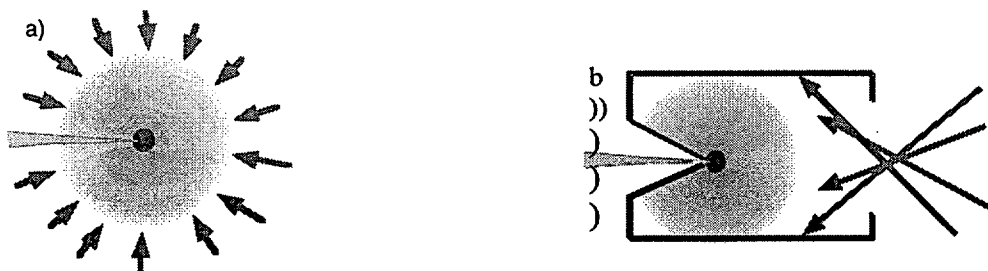


Figure 1.2 Sketch of possible methods of igniting tritium-lean fuels. Sketch a shows a direct-drive system with equal illumination from all angles surrounding target and the ignitor laser from the left. Sketch b shows compression beams entering hohlraum from right, and ignitor laser from left [6,17].

Traditionally, when designing a thick-liquid protected fusion energy chamber such as HYLIFE-II [XX], a major limitation to the choice of the liquid was the tritium-breeding ratio (TBR). The blanket was required to provide a TBR of greater than approximately 1.1 so that tritium did not need to be added to the system during operation. Elimination of this requirement allows for greater flexibility in a thick-liquid selection than ever before. Very little known work has been done suggesting a molten-salt that did not have a requirement of tritium breeding. Materials may now be selected based upon other characteristics, such as: safety and environmental characteristics, pumping power, corrosion, and vapor pressure, along with others.

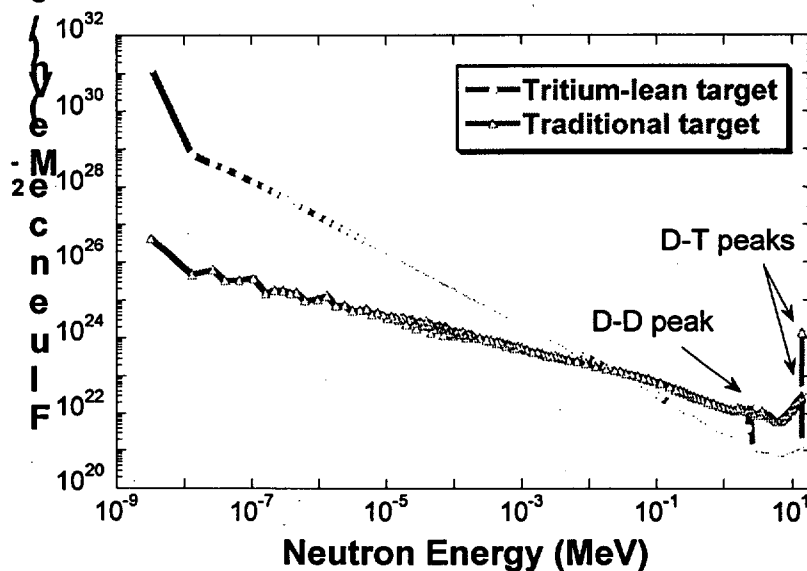


Fig. 1.3 Neutron spectrum emerging from a traditional deuterium-tritium target as well as the tritium-lean target analyzed by Atzeni [4,5].

This study assesses characteristics of single, binary, and ternary molten-salts as well as several liquid metals. Using the National Institute of Standards and Technology (NIST) Properties of Molten Salts Database, approximately 4300 molten-salts are included in the study [1]. Two rounds of analysis were performed and are reported herein. An assessment of the safety and environmental characteristics and a pumping power analysis are both performed on all materials with available density and viscosity data (necessary data for pumping power analysis). Viscosity data is taken for all materials at 900 K.

## 2. Safety and Environmental Assessment

When considering a liquid for use in a fusion chamber, the safety and environmental characteristics are extremely important. Three assessments were done in this study: a calculation of the waste-disposal rating to determine if the material could meet the low-level waste criterion, an analysis of the radioactive afterheat as an indicator of the acceptability of the material from an accident perspective, and a calculation of the contact dose rate to determine if the material could be recycled following its use in the power plant. All analyses assumed a total inventory of material of approximately 1250 m<sup>3</sup>, with approximately 12.5 m<sup>3</sup> (or 1%) of the material in the chamber at any given time. All studies were done using the Monte Carlo code TART and the activation code ACAB [8,9,10]. Neutron irradiation was assumed to occur for 30 full-power years. To simplify the modeling, the neutron spectrum used for neutron activation calculations was that experienced by a 1-cm-thick shell of liquid. We applied this un-attenuated spectrum to the entire liquid blanket to give a model. Analysis was done for all elements on this shell. This conservative assumption was used only for the S&E analyses.

### *Waste-Disposal Rating*

A waste-disposal rating is given to a material in order to classify the method of disposal needed. The waste-disposal rating is given by :

$$WDR = \sum_i \frac{A_i / V}{SAL_i} \quad , \quad (2-1)$$

where  $A_i$  is the activity of the  $i^{\text{th}}$  radionuclide, SAL is its specific activity limit given by Fetter, Cheng, and Mann, and  $V$  is the component volume in m<sup>3</sup> [11].

If the summation of all the radionuclides is taken, a comparison of each component is provided. If the WDR of the component is less than or equal to 1, the material can be disposed of via shallow land burial. Given the potentially large waste volumes involved, disposal via shallow land burial is a primary goal for S&E, and thus, liquids with a WDR greater than unity were eliminated from consideration.

## Radioactive Afterheat

In the case of a severe accident, the radioactive afterheat of the liquid could heat the chamber wall and increase the quantity of material mobilized and released to the environment. We compare the afterheat of the liquid to that from the chamber itself (assumed to be type 304 stainless steel, as in HYLIFE-II).

Results of radioactive afterheat were taken as an integrated afterheat after 7 days. Since only 1% of the liquid is in the chamber at any given time, we compare 1% of the total liquid integrated afterheat to that of the full chamber. Liquids in which the afterheat exceeds that of the chamber were eliminated. According to work done by S. Reyes et al., the temperature of the first wall of the chamber, in an accident scenario, varies only slightly for our time scales. As Figure 2.1 shows, there is a slight peak of temperature at approximately 15 minutes, and from then on (for the first day) the temperature of the wall decreases [12]. Figure 2.1 shows many reactor components, however we are most interested in the first wall result.

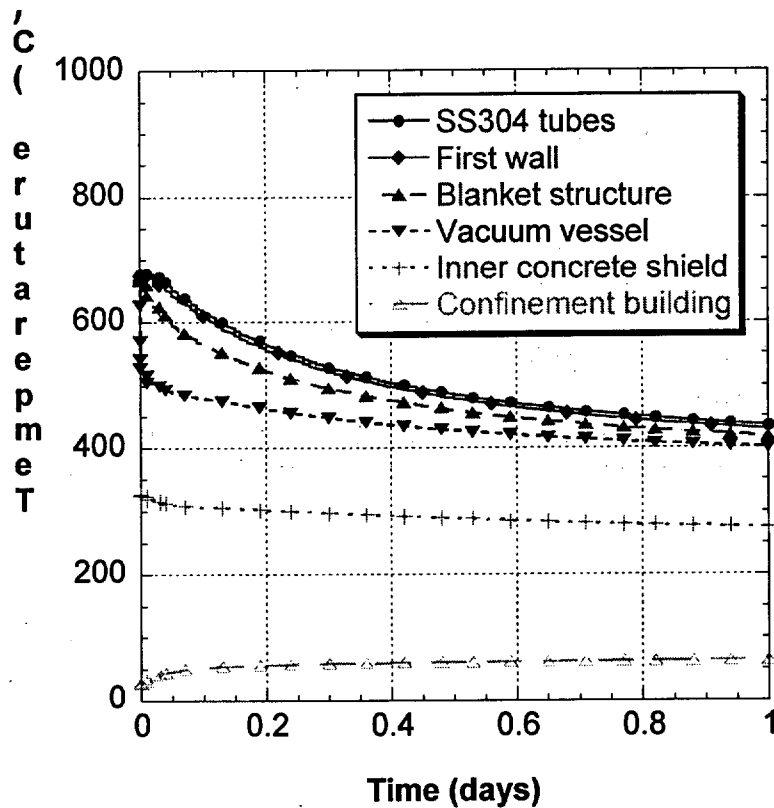


Figure 2.1 Temperature of Reactor Components at different times after accident [12]

In order to keep the SS 304 below its melting temperature ( $T_{\text{melt, SS304}} \sim 1400^\circ \text{C}$ ), our integrated afterheat, at a time of 7 days after accident, must be below  $2.33\text{E}9 \text{ W}$



### *Contact Dose Rate*

The ability to recycle neutron-activated components is another S&E criterion. Specifically, we assume that activated components qualify for remote recycling (e.g., robotic handling) provided that their contact dose limit is  $< 0.1$  Sv/hr following 50 years of radioactive decay. While hands-on recycling is desirable, it requires a significantly lower contact dose rate of  $< 25$   $\mu$ Sv/hr, which may prove to be overly restrictive. Use of the remote recycling assumption allows a larger number of materials to be considered at this early stage of analysis.

### *S&E Results*

A Fortran program was written to input a molten-salt (with the mole percent of each component) and calculate the density of each element in the material. These densities were then compared against the limits (based on S&E criteria) for each element. If all of the elements included in a particular material were present at densities less than or equal to the density limit for that element, then the material passed the S&E screening. Table 2.1 shows the acceptable limits of an element in a molten-salt, and which criterion limits the acceptability. For each material that contained one or more unacceptable elements, the output lists those elements along with the reason for elimination. This helped us better understand the acceptability of each individual element for S&E characteristics. After assessing S&E characteristics, there were approximately 200 liquids remaining—mostly single-salts and binaries. These were then evaluated for pumping power.

Element	Limit (g/cc)	Limiting Factor
Li	1.10E+02	AH
Be	7.53E+03	WDR
B	9.49E+02	WDR
C	8.34E+01	WDR
N	4.78E-02	WDR
O	2.63E+01	WDR
F	1.05E+02	WDR
Ne	1.22E+01	WDR
Na	5.11E+01	CDR
Mg	2.64E+01	AH
Al	3.45E-02	WDR
Si	6.90E+01	WDR
P	3.72E+02	AH
S	2.06E+01	AH
Cl	4.90E-02	WDR
Ar	6.45E-02	WDR
K	5.01E-02	WDR
Ca	1.34E+00	WDR
Sc	5.09E+00	AH
Ti	5.86E+01	AH
V	3.78E+02	AH
Cr	1.41E+03	AH
Mn	1.46E+01	AH
Fe	4.54E+01	CDR
Co	7.13E-04	CDR
Ni	1.02E-01	CDR

Element	Limit (g/cc)	Limiting Factor
Cu	1.85E-01	CDR
Zn	2.29E+01	CDR
Ga	8.48E+00	AH
Ge	1.18E+02	AH
As	2.51E+00	AH
Se	5.51E-02	WDR
Br	1.13E-01	WDR
Kr	2.63E-01	CDR
Rb	3.11E+00	CDR
Sr	7.29E+01	CDR
Y	8.38E+00	AH
Zr	2.77E+00	WDR
Nb	1.81E-05	WDR
Mo	3.32E-04	WDR
Ru	7.41E-03	WDR
Rh	3.54E-02	WDR
Pd	2.05E-03	WDR
Ag	9.04E-05	WDR
Cd	2.88E-02	WDR
In	2.05E+01	AH
Sn	1.63E+01	WDR
Sb	2.00E+00	AH
Te	9.69E-01	WDR
I	2.90E+01	AH
Xe	9.83E-02	CDR
Cs	1.43E-02	CDR

Element	Limit (g/cc)	Limiting Factor
Ba	8.66E-02	CDR
La	1.15E+01	WDR
Ce	1.29E+01	WDR
Pr	3.18E+01	AH
Nd	9.82E-02	CDR
Sm	7.78E-04	CDR
Eu	4.76E-05	CDR
Gd	9.26E-04	WDR
Tb	2.66E-05	WDR
Dy	1.60E-04	WDR
Ho	3.75E-06	WDR
Er	4.64E-04	WDR
Tm	1.35E-02	WDR
Yb	1.64E+01	WDR
Lu	1.49E+01	AH
Hf	1.25E+01	AH
Ta	1.25E+00	AH
W	8.38E+00	WDR
Re	4.93E-01	WDR
Os	6.45E-03	WDR
Ir	9.80E-05	WDR
Pt	7.33E-02	WDR
Au	4.97E+00	AH
Hg	2.04E+02	AH
Tl	3.35E+01	AH
Pb	9.05E+00	WDR
Bi	5.15E-04	WDR

Table 2.1 Maximum density an element can have in a liquid in order to be acceptable for use in thick-liquid protection of the fusion chamber.

*Factor limiting element density:*

*WDR = Waste Disposal Rating, CDR = Contact Dose Rate*

*AH = Radioactive Afterheat*

### 3. Pumping Power Assessment

When considering a molten-salt for thick-liquid protection of a fusion chamber, the pumping power needed to pump the volume of material through the chamber is a very important characteristic in the selection of materials. Pumping power must be sufficiently low in order to maximize net electric power generated from the fusion plant. In the case of HYLIFE-II design, three components to pumping power must be considered: velocity head, frictional/minor losses in pipes, and lifting power.

*Velocity Head*

The molten-salt is delivered via oscillating nozzles in order to achieve a “pocket” for the target to be injected/ignited. This pocket must be thick enough to provide adequate shielding to chamber structure components.

Knowing that an equivalent thickness of flibe (34% BeF<sub>2</sub> – 66% LiF) (56 cm) will provide adequate shielding to the first wall of the chamber, by limiting neutron damage to less than 100 displacements per atom (DPA) after 30 years of continuous irradiation, we determined the thickness of each molten-salt that would result in an equivalent DPA. This equivalent thickness leads to, along with the mean free path (obtained from the Monte Carlo simulation code TART), the number of mean free paths the neutrons would need to be adequately diminished.

In this study, the chamber was assumed to be a spherical shell with inner radius of 0.5 meters. The volume of the liquid blanket can be estimated by the equation:

$$V \propto \frac{4}{3} \cdot \pi \cdot \left[ (R_p + n_s \cdot \lambda_n)^3 - R_p^3 \right] \quad , \quad (3-1)$$

where  $R_p$  is the inner radius of the molten-salt pocket,  $\lambda_n$  is the mean free path of the neutron -- at an energy of 2.539 MeV (mean energy of Atzeni target) -- and  $n_s$  is the number of mean free paths of liquid needed to adequately shield the chamber wall. For HYLIFE-II with traditional deuterium-tritium targets, we require a thickness of 56 cm. This yields a blanket volume of  $\sim 4.5 \text{ m}^3$ .

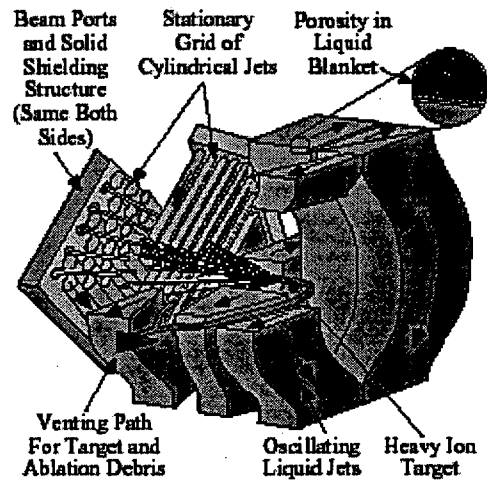
The actual volume of the molten-salt used in the chamber is  $V_c = V\Omega$ , where  $\Omega$  is a geometric factor that adjusts for the complexity of the actual liquid blanket (a spherical shell is only an approximation). Given that the actual volume of flibe in HYLIFE-II is  $12.5 \text{ m}^3$ ,  $\Omega$  is calculated to be 2.8 for our assumed geometry

The height of the blanket is taken to be:

$$h = 2 \cdot (R_p + \lambda_n \cdot n_s) \quad (3-2)$$

Multiplying this height by the repetition rate yields the liquid injection velocity,  $u$ , needed to clear the chamber in time for the next “shot”. This analysis neglects gravitational effects.

Figure 3.1 Depiction of the liquid pocket geometry in HYLIFE-II [13].



Starting from the first principles relation for power, we can derive a relation for the velocity head pumping power:

$$P = E / t$$

$$P = \frac{1}{2} m u^2 \cdot f$$

$$P = \frac{1}{2} \cdot V_c \cdot \rho \cdot u^2 \cdot f ,$$

where  $\rho$  is the liquid density (in kg/m<sup>3</sup>),  $u$  is the liquid velocity (in m/s), and  $f$  is the frequency of shot repetition. Then, from Eqs. (3-1) and (3-2), we can obtain the velocity head pumping power:

$$P = \frac{4}{6} \cdot \pi \cdot \left[ (R_p + n_s \cdot \lambda_n)^3 - R_p^3 \right] \left[ 2 \cdot (R_p + \lambda_n \cdot n_s) \cdot f \right] \cdot \rho \cdot f \quad (3-3)$$

#### *Frictional/Minor Losses*

With a liquid flowing through the piping of the structure, frictional losses are to be expected, and pumping power is needed to overcome these losses. A frictional factor is calculated using the Reynolds number for the flow. The Reynolds number was calculated using the equation [14]:

$$Re = \frac{u \cdot N_d \cdot \rho}{\eta} , \quad (3-4)$$

where  $N_d$  is the thickness of the nozzle (assumed to be 7 cm), and  $\eta$  is the viscosity of the material (in kg/m-s).

For turbulent flow ( $Re > 3000$ ), the friction factor is taken to be [14]:

$$F = \frac{1.325}{\left[ \ln \left( \frac{E_d}{3.7} + \frac{5.74}{Re^{0.9}} \right) \right]^2} , \quad (3-5)$$

where  $E_d$  is the value for the relative roughness of the pipe divided by the diameter of the pipe (in this study  $E_d = 1.5E-5$ ).

For laminar flow ( $Re < 3000$ ), the friction factor is taken to be:

$$F = \frac{64}{Re} \quad (3-6)$$

In order to calculate the pumping power due to frictional losses, an effective piping length is needed. Since our calculations were assuming straight pipes, corrections for bends, joints, etc. were made with an effective length  $(L/D)_{\text{eff}}$ . This number is calculated as:

$$(L/D)_{\text{eff}} = \frac{2 \cdot g \cdot 7.5}{F_{\text{hylife-II}} \cdot Q_{\text{hylife-II}}^2} \quad , \quad (3-7)$$

where  $F_{\text{hylife-II}}$  is the frictional factor of the original HYLIFE-II flibe flow and  $Q_{\text{hylife-II}}$  is the volumetric flow rate of flibe in the original design. For this study  $(L/D)_{\text{eff}} = 361$ .

The pumping power needed to overcome frictional losses in the pipe is described by the equation:

$$P = H \cdot \rho \cdot g \cdot Q \quad , \quad (3-8)$$

where  $H$  is the frictional head loss and is given by Eqn. 3-9.

$$H = \text{head loss} = \frac{\frac{1}{2} \cdot F_f \cdot (L/D)_{\text{eff}} \cdot u_{\text{pipe}}^2}{g} \quad (3-9)$$

The frictional losses for the original HYLIFE-II design, with a traditional 50-50 D-T target, as described by Palmer House are 7.84 MW [15].

Use of the Atzeni target significantly reduces the required flow rate. This is mostly due to the lower repetition rate (1.7 vs. 6.4 Hz), which reduces the liquid velocity. The softer spectrum of the Atzeni target also leads to a thinner pocket (45 cm vs. 56 cm) and the overall frictional losses are only 1.83 MW.

### *Lifting Power*

Pumping power is needed to get the liquid that has been sprayed to the bottom of the chamber back up to the top of the chamber. This pumping power is called lifting power. It is calculated using a 10.5-meter distance from the bottom of the chamber to the top of the nozzle jets. The equation for the lifting power is given by:

$$P = 10.5 \cdot \rho \cdot g \cdot Q \quad (3-10)$$

For the original HYLIFE-II design, the lift power was 10.98 MW. Using the values for the tritium-lean target results in a significant drop in the lifting pumping power to 4.68 MW (for flibe). In this case, the reduction is due entirely to the reduced flow rate.

### *Pumping Power Results*

Sixty-six liquids were analyzed for the total pumping power needed to keep the salt flowing through the chamber at the correct frequency. Acceptable pumping power was assumed to be less than or equal to 80 MW, though the exact value is subject to debate. Nine liquids failed the pumping power requirement. Seven of the nine materials that failed contained boron, an extremely viscous material. The other two materials are  $\text{BeF}_2$  and  $\text{Tl}_2\text{S}$ , both very viscous materials. Materials that fared well in the pumping power assessment usually contained lithium, sodium, or rubidium. Some other materials also pass, but on a less frequent basis. Some typical pumping power results are shown in Table 3.1.

Material #1	BeF2	Hgl2	LiF	LiF	LiF
Material #2	LiF	~	~	NaF	NaF
Material #3	~	~	~	BeF2	BeF2
Mol %1	34.00	100	100	33.3	31.5
Mol %2	66.00	0	0	33.3	31
Mol %3	0.00	0	0	33.4	37.5
Mean Free Path (m)	0.0450	0.1033	0.0349	0.0518	0.0512
Required Thickness (m)	0.45	1.00	0.32	0.51	0.50
Density (kg/m^3)	1974.00	4032	1917	2000	2000
Viscosity (kg/m-s)	9.90E-03	5.04E-04	3.40E-03	1.00E-02	7.80E-02
Velocity Head (MW)	0.15	3.40	0.06	0.21	0.20
Frictional Losses (MW)	1.83	6.93	0.94	2.19	3.61
Lift (MW)	4.68	15.11	3.93	5.05	5.00
Pumping Power Total (MW)	6.66	25.43	4.93	7.45	8.81
Normalized P.P. *	1.00	3.82	0.74	1.12	1.32
Material #1	LiF	LiF	NaBF4	NaF	PbF2
Material #2	NaF	NaF	~	~	~
Material #3	BeF2	~	~	~	~
Mol %1	63	60	100	100	100
Mol %2	5	40	0	0	0
Mol %3	32	0	0	0	0
Mean Free Path (m)	0.0463	0.0544	0.0464	0.0601	0.0401
Required Thickness (m)	0.45	0.49	0.43	0.56	0.44
Density (kg/m^3)	2000	2033	1792	2183	8200
Viscosity (kg/m-s)	1.00E-02	4.19E-03	1.15E-03	4.11E-03	1.00E-03
Velocity Head (MW)	0.15	0.19	0.12	0.29	0.59
Frictional Losses (MW)	1.85	1.74	1.04	2.20	3.80
Lift (MW)	4.75	5.03	4.16	5.75	19.25
Pumping Power Total (MW)	6.75	6.97	5.32	8.24	23.64
Normalized P.P.	1.01	1.05	0.80	1.24	3.55
Material #1	Li17Pb83	Li	Hg	BeF2	B2O3
Material #2	~	~	~	~	~
Material #3	~	~	~	~	~
Mol %1	100	100	100	100.00	100
Mol %2	0	0	0	~	0
Mol %3	0	0	0	~	0
Mean Free Path (m)	0.0494	0.1138	0.0340	0.0513	0.0583
Required Thickness (m)	0.97	1.05	0.47	0.60	0.42
Density (kg/m^3)	9710	530	13530	1959.00	1610
Viscosity (kg/m-s)	2.00E-03	3.43E-04	1.45E-03	2.76E+04	1.57E+02
Velocity Head (MW)	7.38	0.53	1.16	0.33	0.10
Frictional Losses (MW)	17.38	1.17	6.76	454692.41	1809.25
Lift (MW)	35.66	2.05	32.78	5.38	3.70
Pumping Power Total (MW)	60.41	3.75	40.69	454698.13	1813.05
Normalized P.P.	9.07	0.56	6.11	68272.99	272.23

Table 3.1  
Sample of Results of Pumping Power Assessment  
\*Values normalized to flibe (34% BeF<sub>2</sub> – 66% LiF)

#### 4. Conclusions and Recommendations

Upon conclusion of the numerical analysis, approximately 57 liquids passed all evaluations. Most of these salts contain elements such as sodium, lithium, beryllium, boron, fluorine, and oxygen. Other elements were present in lesser frequency. These liquids are presented in Table 4.1.

Molten-Salt Composition			Mol %				Molten-Salt Composition			Mol %		
BeF2	LiF	~	34	66	0		NaPO3	Na2SO4	~	75	25	0
BeF2	LiF	~	50	50	0		NaPO3	Na4P2O7	~	75	25	0
BeF2	LiF	~	75	25	0		NaVO3	~	~	100	0	0
BeF2	NaF	~	30	70	0		NaVO3	V2O5	~	20	80	0
BeF2	NaF	~	50	50	0		NaVO3	V2O5	~	80	20	0
BeF2	RbF	~	50	50	0		Na2CO3	~	~	100	0	0
CaSO4	Na2SO4	~	10	90	0		Na2SO4	~	~	100	0	0
CaSO4	Na2SO4	~	30	70	0		Na2S3	~	~	100	0	0
CaSO4	Na2SO4	~	55	45	0		Na2S4	~	~	100	0	0
FeS	~	~	100	0	0		Na2S5	~	~	100	0	0
HgI2	~	~	100	0	0		Na2WO4	~	~	100	0	0
LiF	~	~	100	0	0		Na4P2O7	~	~	100	0	0
LiF	NaF	BeF2	33.3	33.3	33.4		Na4P2O7	WO3	~	34	66	0
LiF	NaF	BeF2	31.5	31	37.5		Na4P2O7	WO3	~	65	35	0
LiF	NaF	BeF2	63	5	32		RbF	~	~	100	0	0
LiF	NaF	~	60	40	0		RbI	~	~	100	0	0
LiF	RbF	~	43	57	0		Rb2CO3	~	~	100	0	0
LiI	~	~	100	0	0		TiI	~	~	100	0	0
Li2CO3	~	~	100	0	0		V2O5	~	~	100	0	0
Li2CO3	Na2CO3	~	10	90	0		PbF2	~	~	100	0	0
Li2CO3	Na2CO3	~	40	60	0		Rb	~	~	100	0	0
Li2CO3	Na2CO3	~	60	40	0		Li17Pb83	~	~	100	0	0
Li2CO3	Na2CO3	~	90	10	0		Na	~	~	100	0	0
Li2WO4	~	~	100	0	0		Li	~	~	100	0	0
NaBF4	~	~	100	0	0		Hg	~	~	100	0	0
NaBF4	NaF	~	92	8	0		Ga	~	~	100	0	0
NaF	~	~	100	0	0		LiSn	~	~	100	0	0
NaI	~	~	100	0	0		In	~	~	100	0	0
NaPO3	~	~	100	0	0							

Table 4.1 Liquids that passed all assessments.

It is recommended that further analysis be done on these liquids. Further analysis may include corrosion, surface tension, and/or vapor pressure assessments. After additional screening, perhaps 6-12 materials might remain. A detailed analysis of these materials then could be conducted to assess their suitability for use in a thick-liquid, fast ignition inertial confinement fusion energy system.



## 5. Acknowledgements

The authors would like to thank Ralph W. Moir (LLNL) for his help in the overall development of this report. His collaboration has been essential to the completion of the above work. Work was performed under the auspices of the US Department of Energy by University of California Lawrence Livermore National Laboratory under Contract W-7405-Eng-48.

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